

Redetermination and H-atom refinement of
(S)-(+)-ibuprofen. Corrigendum.Lars Kr. Hansen,^{a*} German L.
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Key indicators

Single-crystal X-ray study

T = 298 K

Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$

R factor = 0.038

wR factor = 0.104

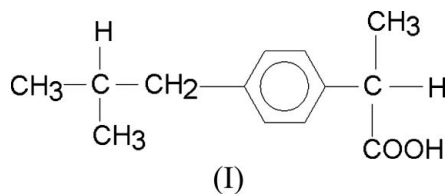
Data-to-parameter ratio = 7.0

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.In the paper by Hansen, Perlovich & Bauer-Brandl [*Acta Cryst.* (2003), E59, o1357–o1358], the coordinates of the *R* enantiomer of the title compound, C₁₃H₁₈O₂, were incorrectly given instead of those of the *S* enantiomer. The correct coordinates of the *S* enantiomer are given here.

Received 8 May 2006

Accepted 31 May 2006

Comment

In the paper by Hansen *et al.* (2003), the coordinates of the *R* enantiomer were incorrectly given instead of those of the *S* enantiomer, (I). The correct coordinates of the *S* enantiomer are given in the deposited replacement CIF. Molecular geometry parameters are not affected, except for the signs of torsion angles; the correct values are given in Table 1 below for the torsion angles in Table 2 of the previous report (where there was also an error in the atom numbering). Fig. 1 shows the correct structure of the two independent molecules, which form a hydrogen-bonded dimer without crystallographic symmetry.

Experimental

Table 1

Selected torsion angles (°).

C5B–C4B–C2B–C3B	29.1 (4)	O1A–C1A–C2A–C4A	–81.7 (4)
C7B–C10B–C11B–C12B	–68.0 (5)	C3A–C2A–C4A–C5A	–144.4 (4)
C4B–C2B–C1B–O1B	83.5 (3)	C7A–C10A–C11A–C13A	67.9 (5)

All H atoms were refined freely [C–H = 0.85 (3)–114 (5) Å].

Data collection: *CAD-4-PC* Software (Enraf–Nonius, 1992); cell refinement: *CELDIM* in *CAD-4-PC* Software; data reduction: *XCAD* (McArdle & Higgins, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEX* (McArdle, 1995); software used to prepare material for publication: *OSCAIL* (McArdle, 1993).

References

Enraf–Nonius (1992). *CAD-4-PC Software*. Enraf–Nonius, Delft, The Netherlands.

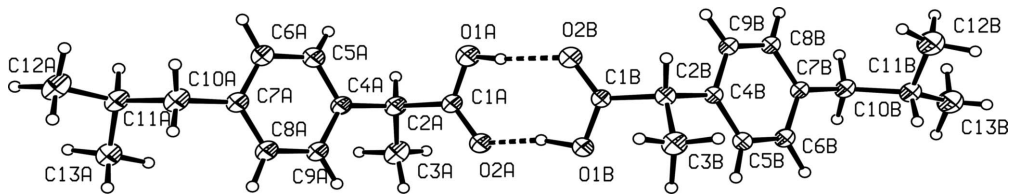


Figure 1

The structure of the two independent molecules of (I). Displacement ellipsoids are shown at the 30% probability level. Hydrogen bonds are shown as dashed lines.

Hansen, L. K., Perlovich, G. L. & Bauer-Brandl, A. (2003). *Acta Cryst.* **E59**, o1357–o1358.

McArdle, P. (1993). *J. Appl. Cryst.* **26**, 752.

McArdle, P. (1995). *J. Appl. Cryst.* **28**, 65.

McArdle, P. & Higgins, T. (1995). *XCAD*. NUI Galway, Ireland.

Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.

Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.

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Key indicators

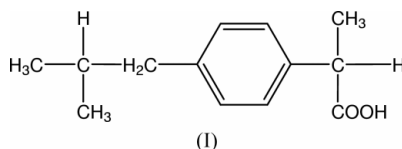
Single-crystal X-ray study
T = 298 K
Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$
R factor = 0.038
wR factor = 0.105
Data-to-parameter ratio = 7.0For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.The crystal structure of (*S*)-(+)-ibuprofen, C₁₃H₁₈O₂, has been redetermined. It crystallizes in the monoclinic space group *P*2₁ with two molecules in the asymmetric unit, giving a cyclic hydrogen-bonded dimer. All the H atoms were located from difference maps and refined isotropically.

Received 20 June 2003

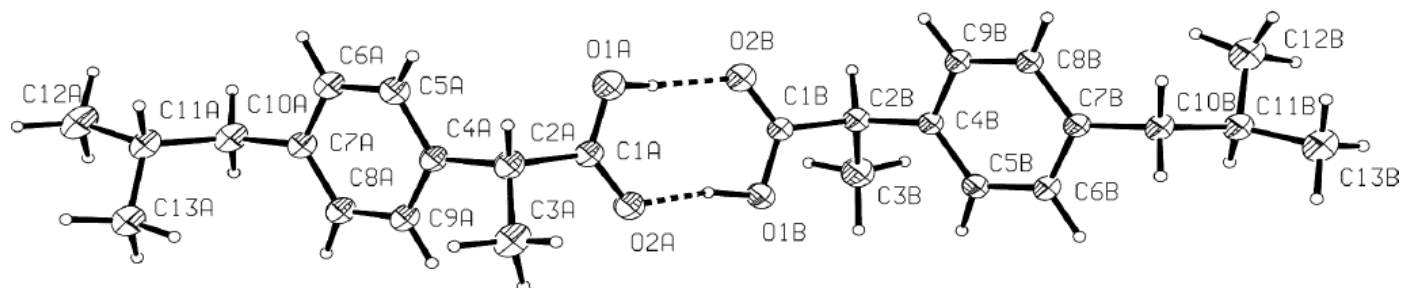
Accepted 30 June 2003

Online 23 August 2003

Comment

The structure of racemic ibuprofen (IBP), (I), has been well described both by X-ray diffraction at 298 K (McConnell, 1974) and single-crystal pulsed neutron diffraction at 100 K (Shankland *et al.*, 1997). The structure of (*S*)-ibuprofen has also been solved by X-ray diffraction (Freer *et al.*, 1993). However, in contrast to the racemate, the quality and paucity of data did not allow the determination of the positions of the H atoms. It should be noted that the lack of this information is a serious obstacle for a logical classification of the hydrogen-bond topology of the compound, and for the calculation of crystal-lattice energy by methods which are sensitive to the geometry of hydrogen bonding (*e.g.* Gavezzotti & Filippini, 1997). With this background, the main aim of this work has been to find and refine the positions of the H atoms.

A view of the two crystallographically independent (*S*)-IBP molecules, with the atomic numbering, is presented in Fig. 1. The parameters of the hydrogen-bond geometry for both the racemic and *S*-enantiomer IBP are shown in Table 1 (Taylor & Kennard, 1982). As there are two independent molecules in the asymmetric unit, the two hydrogen bonds are not geometrically equivalent; one of them is shorter than the other. Comparison of these data gives the following conclusions: (a) the hydrogen bond in the racemate is more linear than in the *S*-enantiomer; (b) one of the O...O distances in (*S*)-IBP is essentially the same as that in the racemate, but the other is longer; (c) the H...O distance in the racemate is an average of the analogous parameters of the *S*-enantiomer. The conformations of the *S*-enantiomer molecules are different; the main parameters characterizing these differences are presented in Table 2, together with corresponding ones for the racemate. As can be seen, molecule *A* in the *S*-enantiomer structure has approximately the same conformation as the IBP molecule in the racemate.


Figure 1

A view of the (*S*)-(+)-ibuprofen dimer, with the atomic numbering scheme. Displacement ellipsoids are drawn at the 20% probability level.

Experimental

(*S*)-IBP single crystals were grown by slow evaporation from an *n*-heptanol solution.

Crystal data

$C_{13}H_{18}O_2$	$D_x = 1.098 \text{ Mg m}^{-3}$
$M_r = 206.27$	Mo $K\alpha$ radiation
Monoclinic, $P2_1$	Cell parameters from 25 reflections
$a = 12.456 (4) \text{ \AA}$	$\theta = 14\text{--}20^\circ$
$b = 8.0362 (11) \text{ \AA}$	$\mu = 0.07 \text{ mm}^{-1}$
$c = 13.533 (3) \text{ \AA}$	$T = 298 (2) \text{ K}$
$\beta = 112.86 (2)^\circ$	Block, colourless
$V = 1248.2 (5) \text{ \AA}^3$	$0.40 \times 0.40 \times 0.30 \text{ mm}$
$Z = 4$	

Data collection

Enraf–Nonius CAD-4 diffractometer	$\theta_{\max} = 27.0^\circ$
ω - 2θ scans	$h = 0 \rightarrow 15$
3086 measured reflections	$k = -1 \rightarrow 10$
2910 independent reflections	$l = -17 \rightarrow 15$
1683 reflections with $I > 2\sigma(I)$	3 standard reflections
$R_{\text{int}} = 0.019$	frequency: 120 min
	intensity decay: 2%

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0583P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.039$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.105$	$(\Delta/\sigma)_{\max} = 0.002$
$S = 0.98$	$\Delta\rho_{\max} = 0.12 \text{ e \AA}^{-3}$
2910 reflections	$\Delta\rho_{\min} = -0.11 \text{ e \AA}^{-3}$
416 parameters	Extinction correction: <i>SHELXL97</i>
All H-atom parameters refined	Extinction coefficient: 0.016 (3)

Table 1

Hydrogen-bond geometry (\AA , $^\circ$) in the crystal structures of the title compound and of its racemate.

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
$O1A\text{--}H1A\cdots O2B$	0.94 (5)	1.73 (6)	2.651 (4)	169 (5)
$O1B\text{--}H1B\cdots O2A$	1.07 (5)	1.58 (5)	2.634 (4)	168 (4)
$O2\text{--}H2O\cdots O1^{1a}$	0.963 (13)	1.664 (10)	2.627 (7)	179.5 (7)

Symmetry code: (i) $1 - x, 1 - y, 1 - z$. Reference: (a) Freer *et al.* (1993).

Table 2

Geometrical and conformational parameters (\AA , $^\circ$) of (*S*)- and racemic IBP.

	(<i>S</i>)-IBP <i>A</i>	(<i>S</i>)-IBP <i>B</i>	racemic IBP
$C5\text{--}C4\text{--}C2\text{--}C3$	144.4 (4)	−29.1 (4)	140.9 (4)
$C7\text{--}C10\text{--}C11\text{--}C12$	−67.9 (5)	68.0 (5)	−67.3 (4)
$C4\text{--}C2\text{--}C1\text{--}O2$	81.7 (4)	−83.5 (3)	88.7 (3)
$O1\text{--}C1$	1.219 (3)	1.226 (3)	1.222 (3)
$O2\text{--}C1$	1.302 (4)	1.302 (4)	1.305 (3)
$C1\text{--}C2$	1.496 (5)	1.518 (4)	1.509 (3)

Data collection: *CAD-4-PC Software* (Enraf–Nonius, 1992); cell refinement: *CELDIM* in *CAD-4-PC Software*; data reduction: *XCAD* (McArdle & Higgins, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEX* (McArdle, 1995); software used to prepare material for publication: *OSCAIL* (Version 9; McArdle, 1993).

This work was generously supported by the Norwegian Research Council (project No. HS 58101).

References

- Enraf–Nonius. (1992). *CAD-4-PC Software*. Version 1.1. Enraf–Nonius, Delft, The Netherlands.
- Freer, A. A., Bunyan, J. M., Shankland, N. & Sheen, D. B. (1993). *Acta Cryst.* **C49**, 1378–1380.
- Gavezzotti, A. & Filippini, G. (1997). *Energetic Aspects of Crystal Packing: Experiment and Computer Simulations*, ch. 3, pp. 61–97, in *Theoretical Aspects and Computer Modeling of the Molecular Solid State*, edited by A. Gavezzotti, p. 237. Chichester: John Wiley and Sons.
- McArdle, P. (1993). *J. Appl. Cryst.* **26**, 752.
- McArdle, P. (1995). *J. Appl. Cryst.* **28**, 65.
- McArdle, P. & Higgins, T. (1995). *XCAD*. National University of Ireland, Galway, Ireland.
- McConnell, J. F. (1974). *Cryst. Struct. Commun.* **3**, 73–75.
- Shankland, N., Wilson, C. C., Florence, A. J. & Cox, P. J. (1997). *Acta Cryst.* **C53**, 951–954.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
- Taylor, R. & Kennard, O. (1982). *J. Am. Chem. Soc.* **104**, 5063–5070.